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mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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Molarity. The number of moles of solute in one liter of solution.
moles of solute. M . liters of solution. 7 . 8 Making a Molar
Solution. Chapter 12 Section 3 Concentration of Solns p. 418-424
. 8 . p. 419. 9. Molarity Sample Problems. You have 3.50 L of
solution that contains 90.0 ...

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Dependence [...]

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$\log K_c = 0.5074$ $K_c = \text{Antilog } 0.5074$ $K_c = 3.22$ Gibb's energy of
cell = $2.895 \text{ kJ mol}^{-1}$ Equilibrium constant of cell = 3.22. RBSE
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Question 1.

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239 CHAPTER 12 SOLUTIONS MANUAL Section 12.3 Liquids and Solids pages 415–424 Section 12.3 Assessment page 424 18. Contrast the arrangement of particles in solids and liquids. The particles are closer together in solids than in liquids because of intermolecular attractions.

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