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51. Write a point of distinction between a metallic solid and an ionic solid other than metallic lustre. Answer: Metallic solids can conduct electricity due to movement of electrons in solid state, whereas ionic solids conduct electricity in molten state or in aqueous solution only due to the movement of ions. 52.

**Important Questions**

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### Question Paper

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**CBSE Test Papers for  
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Check important questions and answers for Class 12 Chemistry Board Exam 2020 from Chapter 1 - The Solid State. These questions are based on the latest CBSE Class 12 Chemistry Syllabus.

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MCQs on Solid State: A Chemistry MCQs test on 'Solid State' with answers, Total Questions: 15. Main Website; ... 4 thoughts on " Solid State MCQs | Question - Paper 1 ... Reply ↓ prashant at . good questions for competative as well as board exams.

## **Solid State MCQs | Question - Paper 1**

If NaCl is doped with

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10-4 mol % of  $\text{SrCl}_2$ ,  
the concentration of  
cation vacancies will be  
( $N_A = 6.023 \times 10^{23}$   
 $\text{mol}^{-1}$ ) (a)  $6.023 \times 10$   
 $15 \text{ mol}^{-1}$  (b)  $6.023 \times 10$   
 $16 \text{ mol}^{-1}$  (c)  $6.023 \times 10$   
 $17 \text{ mol}^{-1}$  (d)  $6.023 \times 10$   
 $14 \text{ mol}^{-1}$

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The Solid State Class  
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Explain how you can determine the atomic mass of an unknown metal if you know its mass density and the dimensions of unit cell of its crystal.

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in line with the NCERT  
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pattern required by the

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- 6.

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